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## 1,4-Bis(3-pyridyloxy)benzene

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#### Abstract

The centroid of the benzene ring of 1,4-bis(3-pyridyloxy)benzene, $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$, lies on a crystallographic center of symmetry. The pyridine rings lie trans to each other at an angle of $66.9(1)^{\circ}$ to the central benzene ring. $\pi$-Stacking interactions between the benzene and pyridine rings in adjacent molecules lead to a sinusoidal packing motif along the $b$ axis.


## Comment

We are currently studying certain metallocyclic transition metal complexes containing pyridine donor moieties in order to explore their potential as hosts for small molecules. The ability of some multinuclear palladium(II) complexes to bind small aromatic molecules has been demonstrated recently (Fujita \& Ogura, 1996). We are interested in extending these studies to species containing various other transition metal centers. As the binding of potential guests will be promoted, at least in part, by $\pi$-stacking interactions, it was of interest to examine the nature of any stacking interactions in the free ligands. We have previously reported the preparation of a novel binuclear silver complex of 1,4-bis(2pyridyloxy)benzene (Hartshorn \& Steel, 1996), and have now prepared an isomeric ligand, 1,4-bis(3-pyridyloxy)benzene, (1), by an Ullman reaction between 3-hydroxypyridine and 1,4-dibromobenzene. In order to confirm the structure of this new ligand and to examine its conformation and molecular packing, a single-crystal X-ray determination was carried out.

(1)

The structure of the title molecule, which lies on a crystallographic center of inversion, is shown in Fig. 1. The pyridine rings lie trans to each other, with the dihedral angle between the pyridine and benzene rings being $66.9(1)^{\circ}$. The torsion angles about the O atom are -13.5 (2) [C3i- $\mathrm{Cl}-\mathrm{Ol}-\mathrm{C}^{\prime}$; symmetry code: (i) $2-x,-y, 1-z]$ and $-60.60(17)^{\circ}(\mathrm{C} 1-$ $\mathrm{O} 1-\mathrm{C}^{\prime}-\mathrm{C}^{\prime}$ ). The structure of (1) is similar to those of related 1,4-diheteroaryloxybenzenes (Drendel
\& Sundaralingam, 1985; Katritzky et al., 1996). The fact that the pyridine rings are not orthogonal to the benzene ring, thereby minimizing steric interactions [as observed in 1,4-diphenoxybenzene (Clayden et al., 1990)], suggests a degree of conjugation between the rings mediated by the ether O atoms (Drendel \& Sundaralingam, 1985).


Fig. 1. Perspective view and atom labeling of the title ligand, (1). Displacement ellipsoids are drawn at the $50 \%$ probability level and H atoms are drawn as small circles of arbitrary radii. Symmetry code as in Table 1.

The molecules are arranged in a sinusoidal manner along the $b$ axis as shown in Fig. 2, with the packing being controlled by a $\pi$-stacking interaction between the benzene ring of one molecule and the pyridine rings of two adjacent molecules. The distance between mean planes is 3.576 (2) $\AA$ and there is a slight offset of the centroids of the rings. Stacking interactions between pairs of benzene rings or pairs of pyridine rings are not observed, consistent with the electrostatic nature of the $\pi$-stacking interaction (Hunter \& Sanders, 1990).


Fig. 2. The molecular packing diagram of (1).

## Experimental

1,4-Bis(3-pyridyloxy)benzene, (1), was prepared by a modification of literature procedures (Pavia et al., 1988; Bacon \& Stewart, 1965). 3-Hydroxypyridine ( 32 mmol ). 1,4-dibromobenzene ( 32 mmol ), $\mathrm{K}_{2} \mathrm{CO}_{3}(33 \mathrm{mmol})$ and $\mathrm{Cu}^{0}(2.5 \mathrm{~mol} \%)$ were refluxed in dimethylacetamide ( 10 ml ) under argon for 18 h . The reaction mixture was then diluted with diethyl ether, washed with dilute aqueous NaOH , water and brine, concentrated and chromatographed on silica (ethyl acetate-petroleum ether $1: 1$ ) to give the product as a light-brown solid ( $17 \%$, m.p. $353-354 \mathrm{~K}$ ). Analysis: $\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$ requires C 72.72 . H $4.58, \mathrm{~N} 10.60 \%$; found C $72.46, \mathrm{H} 4.47$. N $10.52 \%$. ${ }^{1} \mathrm{H}$

NMR ( $\left.\mathrm{CDCl}_{3}, 300 \mathrm{MHz}, \mathrm{TMS}\right) ~ \delta=7.05(s, 4 \mathrm{H}, \mathrm{H} 2,3,5,6)$, $7.29\left(m, 4 \mathrm{H}, \mathrm{H} 4^{\prime}, 5^{\prime}\right), 8.37$ ( $d d, 2 \mathrm{H}, J=1.8,3.9 \mathrm{~Hz}, \mathrm{H}^{\prime}$ ), 8.41 $\left(t, 2 \mathrm{H}, J=1.5 \mathrm{~Hz}, \mathrm{H}^{\prime}\right)$. Recrystallization from ethyl acetate gave a mass of crystals from which a suitable crystal was cut.

## Crystal data

$\mathrm{C}_{16} \mathrm{H}_{12} \mathrm{~N}_{2} \mathrm{O}_{2}$
$M_{r}=264.28$
Monoclinic
$P 2_{1} / c$
$a=6.9174$ (4) $\AA$
$b=7.4267$ (4) $\AA$
$c=12.0693(7) \AA$
$\beta=98.206(3)^{\circ}$
$V=613.69(6) \AA^{3}$
$Z=2$
$D_{x}=1.430 \mathrm{Mg} \mathrm{m}^{-3}$
$D_{m}$ not measured

## Data collection

Siemens SMART CCD diffractometer
Exposures over $0.3^{\circ} \varphi$ or $\omega$ rotation scans
Absorption correction: none
1988 measured reflections
903 independent reflections
823 reflections with
$I>2 \sigma(I)$

Mo $K \alpha$ radiation
$\lambda=0.71073 \AA$
Cell parameters from 2080 reflections
$\theta=3.0-26.9^{\circ}$
$\mu=0.096 \mathrm{~mm}^{-1}$
$T=158$ (2) K
Block
$0.48 \times 0.44 \times 0.32 \mathrm{~mm}$
Beige
$R_{\text {int }}=0.047$
$\theta_{\text {max }}=25.85^{\circ}$
$h=-3 \rightarrow 8$
$k=-8 \rightarrow 8$
$l=-9 \rightarrow 14$
No standard reflections
Intensity decay: $10 \%$

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.037$
$(\Delta / \sigma)_{\text {max }}<0.001$
$\Delta \rho_{\text {max }}=0.175 \mathrm{e}^{-3}$
$w R\left(F^{2}\right)=0.105$
$S=1.09$
903 reflections
91 parameters
$H$ atoms not refined
$w=1 /\left[\sigma^{2}\left(F_{0}^{2}\right)+(0.0549 P)^{2}\right.$
$+0.1532 P]$
where $P=\left(F_{o}^{2}+2 F_{c}^{2}\right) / 3$ $\Delta \rho_{\text {min }}=-0.178 \mathrm{e}^{-3}$
Extinction correction: none Scattering factors from International Tables for Crystallography (Vol. C)

Table 1. Selected geometric parameters $\left(\AA,^{\circ}\right)$

| C1--C2 | 1.379 (2) | $\mathrm{N} 1^{\prime}-\mathrm{C} 2{ }^{\prime}$ | 1.339 (2) |
| :---: | :---: | :---: | :---: |
| $\mathrm{Cl}-\mathrm{C}^{1}$ | 1.383 (2) | $\mathrm{C} 2^{\prime}-\mathrm{C} 3^{\prime}$ | 1.381 (2) |
| $\mathrm{Cl} \ldots \mathrm{O} 1$ | 1.399 (2) | $\mathrm{C} 3^{\prime}-\mathrm{C} 4^{\prime}$ | 1.384 (2) |
| C2--C3 | 1.383 (2) | $\mathrm{C} 4{ }^{\prime}-\mathrm{C} 5^{\prime}$ | 1.381 (2) |
| $\mathrm{Ol}-\mathrm{Cl}^{\prime}$ | 1.380 (2) | C5 ${ }^{\prime}-\mathrm{C} 6^{\prime}$ | 1.382 (2) |
| $\mathrm{N} \mathrm{l}^{\prime}-\mathrm{Cb}^{\prime}$ | 1.334 (2) |  |  |
| C2-C1-C3 ${ }^{\text {i }}$ | 120.05 (13) | $\mathrm{N1}{ }^{\prime}-\mathrm{C} 2^{\prime}-\mathrm{C} 3^{\prime}$ | 122.94 (13) |
| $\mathrm{C} 2-\mathrm{Cl}-\mathrm{Ol}$ | 115.98 (12) | $\mathrm{O} 1-\mathrm{C} 3^{\prime}-\mathrm{C} 2^{\prime}$ | 117.49 (13) |
| C3'-C1-O1 | 123.94 (13) | $\mathrm{Ol}-\mathrm{C3}^{\prime}-\mathrm{C4}^{\prime}$ | 122.73 (12) |
| $\mathrm{C} 1-\mathrm{C} 2-\mathrm{C} 3$ | 120.21 (13) | $\mathrm{C} 2^{\prime}-\mathrm{C3}^{\prime}-\mathrm{C}^{\prime}{ }^{\prime}$ | 119.69 (13) |
| $\mathrm{Cl}{ }^{\text {- }} \mathrm{C} 3-\mathrm{C} 2$ | 119.75 (13) | $\mathrm{C} 5^{\prime}-\mathrm{C} 4^{\prime}-\mathrm{Cl}^{\prime}$ | 117.60 (13) |
| C3'-OI-Cl | 118.18 (10) | $\mathrm{C} 4^{\prime}-\mathrm{C} 5^{\prime}-\mathrm{C} 6^{\prime}$ | 119.09 (13) |
| $\mathrm{C} 6^{\prime}-\mathrm{Nl}^{\prime}-\mathrm{C} 2^{\prime}$ | 116.90 (12) | $\mathrm{Na}^{\prime}-\mathrm{C}^{\prime}-\mathrm{C5}^{\prime}$ | 123.73 (14) |

Symmetry code: (i) $2-x,-y, 1-z$.
Crystal decay was monitored by measurement of duplicate reflections.

Data collection: SMART (Siemens, 1995). Cell refinement: SAINT. Data reduction: SAINT (Siemens, 1995). Program(s) used to solve structure: SHELXS86 (Sheldrick, 1990). Pro-
gram(s) used to refine structure: SHELXL93 (Sheldrick, 1993). Molecular graphics: SHELXL93. Software used to prepare material for publication: SHELXL93.

Supplementary data for this paper are available from the IUCr electronic archives (Reference: TA1207). Services for accessing these data are described at the back of the journal.

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# $(-)-(2 S)$-Tifluadom Hydrochloride and Two of its (-)-(2S)-2-(Acylaminomethyl)benzodiazepine Hydrochloride Analogues 

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## Abstract

The structures of three ( - )-( $2 S$ )-(acylaminomethyl)benzodiazepine hydrochlorides, namely, 5-(2-fluoro-phenyl)-1-methyl-2-(3-thenoylaminomethyl)-2,3-dihydro1 H -1,4-benzodiazepin-4-ium chloride $\left[\mathrm{C}_{22} \mathrm{H}_{21} \mathrm{FN}_{3} \mathrm{OS}^{+}\right.$.-$\mathrm{Cl}^{-}$, (1), tifluadom], 1,7-dimethyl-5-phenyl-2-(3-thenoyl-aminomethyl)-2,3-dihydro-1 H -1,4-benzodiazepin-4-ium chloride $\left[\mathrm{C}_{23} \mathrm{H}_{24} \mathrm{~N}_{3} \mathrm{OS}^{+} . \mathrm{Cl}^{-}\right.$, (2)] and 5-(4-fluoro-phenyl)-1,8-dimethyl-2-( $p$-toluoylaminomethyl)-2,3-

